Contents lists available at ScienceDirect

**Tetrahedron Letters** 

journal homepage: www.elsevier.com/locate/tetlet

# Synthesis of carbadisaccharide mimics of galactofuranosides

# Jens Frigell, Ian Cumpstey \*

Department of Organic Chemistry, Arrhenius Laboratory, Stockholm University, 106 91 Stockholm, Sweden

### ARTICLE INFO

### ABSTRACT

charides with excellent regioselectivity.

Article history: Received 20 May 2009 Revised 17 June 2009 Accepted 26 June 2009 Available online 1 July 2009

Keywords: Carbasugars Galactofuranose Pseudodisaccharides Ethers

Galactose is found in the unusual furanose configuration in bacteria and other lower organisms, some of which are pathogenic, but not in mammals.<sup>1</sup> The arabinogalactan from the cell wall of *Mycobacterium tuberculosis* contains a polymeric region of galactofuranosides with alternating ( $\beta$ 1 $\rightarrow$ 5)- and ( $\beta$ 1 $\rightarrow$ 6)-linkages, anchored to the cell-wall peptidoglycan by a ( $\beta$ 1 $\rightarrow$ 4)-linkage to rhamnose.<sup>2</sup> Hydrolytically stable mimics of fragments of this oligosaccharide may be of interest to investigate the substrate-binding properties of the galactosyltransferases<sup>3–5</sup> that assemble the cell wall polysaccharide and could be interesting targets for inhibition. Carbasugar<sup>6,7</sup> pseudodisaccharides have been previously shown to act as glycosyltransferase substrates.<sup>8</sup>

We recently reported the synthesis of a carbagalactofuranose monomer **5**.<sup>9</sup> For the synthesis of carbasugar-containing pseudodisaccharides mimicking the Galf( $\beta 1 \rightarrow 5$ )Galf and Galf( $\beta 1 \rightarrow 6$ )Galf linkages of arabinogalactan, the ether linkages should ideally be constructed in a stereocontrolled manner, and we therefore considered S<sub>N</sub>2 type processes. A versatile approach would be to use a carbagalactofuranose C-1 electrophile that could be attacked by OH-6 or OH-5 carbasugar nucleophiles, or even other alcohols should it be desirable to synthesise other carbagalactofuranosides. 1,2-Epoxides derived from carbapyranoses have been used as electrophiles for the synthesis of pseudodisaccharides with alcohol nucleophiles under Lewis acidic or basic conditions.<sup>10</sup> In this Letter, we report the extension of this concept to the five-membered ring system, and describe the synthesis of some carbagalactofuranoside-containing pseudodisaccharides. Diol  $1^9$  was converted into the epoxide **2** as follows: treatment with tosyl chloride and pyridine gave the 1-tosylate as the major product (42%) along with minor amounts of 2-tosylate (14%) and 1,2-ditosylate (5%). The orientation of substitution was determined by NMR spectroscopy (coupling between the OH proton and either H-1 or H-2). Treatment of the major regioisomer with sodium hydride eliminated tosylate to give the required  $\alpha$ -galacto epoxide **2** (87%). The same epoxide **2** was better obtained in one step as a single diastereomer (87%) from the diol **1** under Mitsunobu conditions (Scheme 1).

BnO

BnO

A partially protected carbagalactofuranose was converted into a 1,2-anhydro derivative. This epoxide was

opened with alcohol nucleophiles under Lewis acid catalysis to give β-carbagalactofuranose pseudodisac-



**Scheme 1.** Reagents and conditions: (a) DIAD, PPh<sub>3</sub>, THF, 0 °C, 86%; (b) (i) BnBr, NaH, DMF, 92%; (ii) ZnCl<sub>2</sub>, Ac<sub>2</sub>O, AcOH, 76%; (iii) NaOMe, MeOH, rt, quant.; (c) (i) acetone, CSA; (ii) BnBr, NaH, DMF; (iii) AcOH, H<sub>2</sub>O, 56% (three steps); (iv) Bu<sub>2</sub>SnO, MeOH, 60 °C; (v) BnBr, CsF, DMF, 88% (two steps).







<sup>\*</sup> Corresponding author. Tel.: +46 0 8 16 2481; fax: +46 0 8 15 4908. *E-mail address*: cumpstey@organ.su.se (I. Cumpstey).

<sup>0040-4039/\$ -</sup> see front matter  $\odot$  2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2009.06.115



Scheme 2. Reagents and conditions: (a) ROH (equiv: a-c, 10; d, 3.9; e, 3.0; f, 3.2; g, h, 5.0), BF3.OEt2 (0.1 equiv), CH2Cl2, rt; (b) Ac2O, pyridine, DMAP, rt.

The required OH-6 and OH-5 alcohols (**3** and **4**) were prepared as follows: perbenzylation of the carbasugar **1** was followed by selective acetolysis of the O-6 benzyl ether and deacetylation to give the primary alcohol **3**. From the carbasugar **5**,<sup>9</sup> OH-5 and OH-6 were protected as an isopropylidene acetal. The remaining three hydroxy groups were benzylated, and the isopropylidene protection was removed to give the 5,6-diol, which was selectively protected at C-6 to give the secondary alcohol **4** (Scheme 1).

Next, we attempted the etherification reaction, first using model alcohols to open the epoxide **2** using BF<sub>3</sub>·OEt<sub>2</sub> as promoter. Ethanol, isopropanol and *tert*-butanol (10 equiv) each gave a single regioisomer **6a–c** of the respective ethers in good yield. The regioselectivity of the reaction was confirmed by acetylation of the products to give acetates **7a–c**; OH-2 was acetylated, as was evident from the downfield shift of H-2 in the <sup>1</sup>H NMR spectra (Scheme 2). Our assignment of the stereochemistry of the epoxide **2** (and thence the products of ring-opening **6**) was confirmed by ethylation of both the diol **1** and the alcohol **6a** obtained from opening of the epoxide **2** by ethanol. The two diethyl derivatives **8** obtained were identical with one another (Scheme 3).

The primary and secondary carbasugar alcohols **3** and **4** both opened the epoxide **2** under the same Lewis acid catalysed conditions with essentially complete regioselectivity for attack at C-1 to



Scheme 3. Reagents and conditions: (a) NaH, EtBr (10 equiv), DMF, rt; 37% from 6a; 45% from 1.

give bis-carbadisaccharides **6d**, **e** as the only pseudodisaccharide products (Scheme 2).<sup>11</sup> Primary and secondary pyranoid carbohydrate alcohols (Rha O-4,<sup>12</sup> Man O-3<sup>13</sup> and Man O-6<sup>14</sup>) also opened the epoxide with complete regioselectivity to give pseudodisaccharide products **6f–h** resembling substructures of other bacterial polysaccharides. That the sense of regioselectivity was the same for the carbasugar and carbohydrate nucleophiles as for the simple alcohols was confirmed by acetylation. By-products with mass spectral data consistent with the pseudotrisaccharides arising from attack of the product alcohols **6** on the epoxide **2** were also seen.

The explanation of the excellent regioselectivity in the epoxide opening may be both steric and electronic in origin. C-1 is expected to be less hindered than C-2 as the C-4a methylene group (flanking C-1) is smaller than the corresponding benzyl-ether-substituted C-3 (flanking C-2). Attack at C-1 leads to the all *trans*  $\beta$ -galacto configuration, while attack at C-2 would give an  $\alpha$ -talo configuration with a 2,3-*cis* relationship. Under Lewis acid catalysed epoxide opening, attack will usually occur at the carbon most able to stabilise a partial positive charge. The more electron-withdrawing nature of the oxygenated C-3 compared to the methylene C-4a is also expected to favour attack at C-1 over C-2.

To conclude, we have synthesised carbafuranoside pseudodisaccharides for the first time. The regioselective Lewis acid catalysed epoxide opening gives the ether-linked pseudodisaccharides via attack at C-1. Pseudodisaccharide mimics of all three galactofuranoside linkages in mycobacterial arabinogalactan are accessible by this method.

#### Acknowledgements

We gratefully acknowledge financial support from the Swedish Research Council (Vetenskapsrådet) and an Ivar Bendixson grant.

## Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2009.06.115.

#### **References and Notes**

- 1. Peltier, P.; Euzen, R.; Daniellou, R.; Nugier-Chauvin, C.; Ferrieres, V. Carbohydr. Res. 2008, 343, 1897.
- 2. Mahapatra, S.; Basu, J.; Brennan, P. J.; Crick, D. C. In Tuberculosis and the Tubercle Bacillus; Cole, S. T., Eisenach, K. D., McMurray, D. N., Jacobs, W. R., Jr., Eds.; American Society for Microbiology: Washington, DC, 2005; p 275
- 3. Belanova, M.; Dianiskova, P.; Brennan, P. J.; Completo, G. C.; Rose, N. L.; Lowary, T. L.; Mikusova, K. J. Bacteriol. 2008, 190, 1141.
- 4. Completo, G. C.; Lowary, T. L. J. Org. Chem. **2008**, 73, 4513.
- Rose, N. L.; Completo, G. C.; Lin, S.-J.; McNeil, M.; Palcic, M. M.; Lowary, T. L. J. 5. Am. Chem. Soc. 2006, 128, 6721.
- McCasland, G. E.; Furuta, S.; Durham, L. J. J. Org. Chem. **1966**, 31, 1516. Ogawa, S. Trends Glycosci. Glyc. **2004**, 16, 33. 6.
- 7 8.
- Ogawa, S.; Matsunaga, N.; Palcic, M. *Carbohydr. Lett.* **1997**, *2*, 299. 9. Frigell, J.; Cumpstey, I. Tetrahedron Lett. 2007, 48, 9073.
- 10. Ogawa, S.; Sasaki, S.-I.; Tsunoda, H. Chem. Lett. **1993**, 1587.

- 11. Representative procedure for epoxide-opening; synthesis of 6d: Epoxide 2 (31 mg, 0.072 mmol) and alcohol 3 (150 mg, 0.28 mmol) were dissolved in  $CH_2Cl_2$  (0.75 mL) under N<sub>2</sub> at rt. BF<sub>3</sub>·OEt<sub>2</sub> (18 µL, 0.14 mmol) was dissolved in  $CH_2Cl_2$  (2.5 mL), and 125  $\mu$ L (7  $\mu$ mol) of this solution was added to the reaction mixture, which instantly turned from colourless to pale yellow. After 10 min, TLC (toluene/EtOAc, 5:1) showed complete consumption of epoxide 2 ( $R_{\rm f}$  0.8), remaining alcohol 3 ( $R_{\rm f}$  0.4) and the formation of a product ( $R_f$  0.5). The reaction was quenched by addition of Et<sub>3</sub>N (0.5 mL) and the mixture was concentrated in vacuo. The crude product was purified by column chromatography on silica gel (toluene/ EtOAc, 4:1) to give the pseudodisaccharide **6d** (48 mg, 69%) as a colourless oil.
- 12. Liptak, A.; Imre, J.; Nanasi, P. Carbohydr. Res. 1981, 92, 154.
- 13. Ennis, S. C.; Cumpstey, I.; Fairbanks, A. J.; Butters, T. D.; Mackeen, M.; Wormald, M. R. Tetrahedron 2002, 58, 9403.
- 14. Ruiz Contreras, R.; Kamerling, J. P.; Vliegenthart, J. F. G. Recl. Trav. Chim. Pays-Bas 1991, 110, 85.